

7-Benzenesulfonamido-3-methyl-8-oxo-5-thia-1-azabicyclo[4.2.0]octa-2-ene-2-carboxylic acid monohydrate

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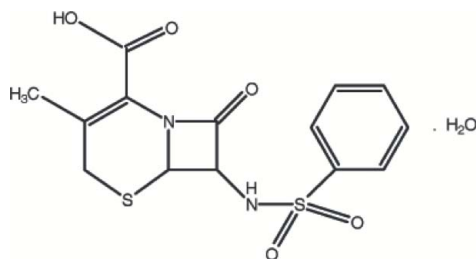
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Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.007$ Å; R factor = 0.055; wR factor = 0.109; data-to-parameter ratio = 15.8.

In the title compound, $\text{C}_{14}\text{H}_{14}\text{N}_2\text{O}_5\text{S}_2 \cdot \text{H}_2\text{O}$, the six-membered ring fused to the β -lactam unit has a twisted conformation. Weak intramolecular $\text{N}-\text{H} \cdots \text{S}$ and $\text{C}-\text{H} \cdots \text{O}$ interactions occur. Intermolecular $\text{C}-\text{H} \cdots \text{S}$, $\text{N}-\text{H} \cdots \text{O}$, $\text{C}-\text{H} \cdots \text{O}$ and $\text{O}-\text{H} \cdots \text{O}$ hydrogen-bonding interactions stabilize the crystal structure, forming a three-dimensional network. Weak $\text{C}-\text{H} \cdots \pi$ interactions are also present.

Related literature

For the production of 7-amino-deacetoxycephalosporanic acid-like components by direct fermentation, see: Schroen *et al.* (2000). For 7-benzenesulfonamido-3-ethenyl-8-oxo-5-thia-1-azabicyclo[4.2.0]octa-2-ene-2-carboxylic acid methanol solvate, see: Mariam *et al.* (2009). For structures with the β -lactam unit, see: Akkurt *et al.* (2008*a,b,c*); Baktır *et al.* (2009); Pınar *et al.* (2006); Yalçın *et al.* (2009); Çelik *et al.* (2009*a,b*). For puckering parameters, see: Cremer & Pople (1975).



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Experimental

Crystal data

$\text{C}_{14}\text{H}_{14}\text{N}_2\text{O}_5\text{S}_2 \cdot \text{H}_2\text{O}$
 $M_r = 372.43$
Orthorhombic, $P2_12_12_1$
 $a = 5.9535$ (7) Å
 $b = 15.8248$ (19) Å
 $c = 18.411$ (2) Å

$V = 1734.6$ (3) Å³
 $Z = 4$
Mo $K\alpha$ radiation
 $\mu = 0.34$ mm⁻¹
 $T = 296$ K
 $0.20 \times 0.20 \times 0.20$ mm

Data collection

Bruker Kappa APEXII CCD area-detector diffractometer
10847 measured reflections

3572 independent reflections
1890 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.064$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.055$
 $wR(F^2) = 0.109$
 $S = 0.97$
3572 reflections
226 parameters
3 restraints

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.20$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.25$ e Å⁻³
Absolute structure: Flack (1983), 1425 Friedel pairs
Flack parameter: -0.07 (11)

Table 1

Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$\text{O1}-\text{H1} \cdots \text{OW1}^i$	0.82	1.86	2.663 (6)	166
$\text{OW1}-\text{HW1} \cdots \text{O5}^{ii}$	0.83 (4)	2.14 (5)	2.799 (6)	136 (5)
$\text{N2}-\text{H2} \cdots \text{S1}$	0.86	2.82	3.136 (3)	103
$\text{N2}-\text{H2} \cdots \text{O2}^{iii}$	0.86	2.30	2.846 (4)	122
$\text{OW1}-\text{HW2} \cdots \text{O3}$	0.84 (5)	2.54 (5)	3.135 (6)	129 (5)
$\text{C3}-\text{H3A} \cdots \text{O1}$	0.96	2.22	2.902 (6)	127
$\text{C6}-\text{H6} \cdots \text{S1}^{iv}$	0.98	2.80	3.756 (4)	165
$\text{C8}-\text{H8} \cdots \text{O4}^i$	0.98	2.30	3.100 (5)	138
$\text{C11}-\text{H11} \cdots \text{O3}^v$	0.93	2.51	3.192 (6)	130
$\text{C3}-\text{H3B} \cdots \text{Cg3}^{ii}$	0.96	2.72	3.640 (5)	161

Symmetry codes: (i) $x-1, y, z$; (ii) $-x+\frac{3}{2}, -y+1, z+\frac{1}{2}$; (iii) $x+1, y, z$; (iv) $x-\frac{1}{2}, -y+\frac{1}{2}, -z$; (v) $x+\frac{1}{2}, -y+\frac{3}{2}, -z$. Cg3 is the centroid of the C9–C14 ring.

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINTE* (Bruker, 2007); data reduction: *SAINTE*; program(s) used to solve structure: *SIR97* (Altomare *et al.*, 1999); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999) and *PLATON* (Spek, 2009).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: VM2014).

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supplementary materials

Acta Cryst. (2010). E66, o73-o74 [doi:10.1107/S1600536809051769]

7-Benzenesulfonamido-3-methyl-8-oxo-5-thia-1-azabicyclo[4.2.0]octa-2-ene-2-carboxylic acid monohydrate

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Comment

One of the building blocks of cephalosporin antibiotics is 7-amino-deacetoxycephalosporanic acid (7-ADCA). It is currently produced from penicillin G using an elaborate chemical ring-expansion step followed by an enzyme-catalyzed hydrolysis. However, 7-ADCA-like components can also be produced by direct fermentation (Schroen *et al.* 2000).

In the title molecule (I) shown in Fig. 1, the β -lactam unit (N1/C6–C8) has a twisted conformation with the dihedral angles of 164.7 (4)° and 164.7 (4)° between the planes N1 C6 C7 and C6 C8 C7 and the planes N1 C6 C8 and N1 C8 C7, respectively. The six-membered ring fused to the β -lactam unit, (N1/S1/C1/C2/C4/C6) is puckered with the puckering parameters (Cremer & Pople, 1975): $Q_T = 0.618$ (3) Å, $\theta = 54.2$ (4)°, $\varphi = 340.4$ (5)°, respectively.

The crystal structure is stabilized by intermolecular C—H \cdots S, N—H \cdots O, C—H \cdots O and O—H \cdots O hydrogen bonding interactions between symmetry-related molecules, forming a network in three dimensions (Table 1, Fig. 2). Furthermore, there is a weak C—H \cdots π interaction [C3—H3B \cdots Cg3(-x + 3/2, -y + 1, z + 1/2); H3B \cdots Cg3 = 2.72 Å, C3 \cdots Cg3 = 3.640 (5) Å, C3—H3B \cdots Cg3 = 161°, where Cg3 is a centroid of the phenyl ring C9–C14].

Experimental

7-ADCA (1 g, 4.7 mmol) was dissolved in distilled water (20 ml) in a round bottom flask (50 ml). Na₂CO₃ (3M) solution was added to maintain the solution at pH 8–9, then benzene sulfonyl chloride (0.82 g, 4.7 mmol) was added dropwise to the solution and stirred at room temperature. As all benzene sulfonyl chloride was consumed, pH becomes stable at 8–9, which confirms the completion of reaction. Then pH was adjusted to 1–2, by using 3 N HCl. The precipitate obtained was filtered, washed with distilled water, dried and recrystallized in ethyl acetate. Light yellow prisms of (I) appeared after two days.

Refinement

The H atoms of the water molecule were located in difference Fourier maps and were refined with O—H distances restrained to 0.83 (1) Å and H \cdots H distances restrained to 1.30 (1) Å, with displacement parameters fixed at 1.5 times U_{eq} of the parent O atoms. H atom on O1 was calculated and refined with a riding model [O—H = 0.82 Å and $U_{iso}(H) = 1.5U_{eq}(O)$]. The other H atoms were placed geometrically, with N—H = 0.86 Å, C—H = 0.96–0.98 Å, and included in the refinement using a riding model, with $U_{iso}(H) = 1.2$ or $1.5U_{eq}(\text{parent atom})$.

Figures

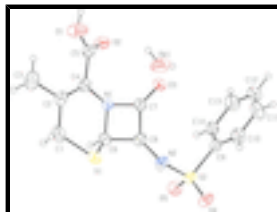


Fig. 1. The title molecule with the atom numbering scheme. Displacement ellipsoids for non-H atoms are drawn at the 30% probability level.

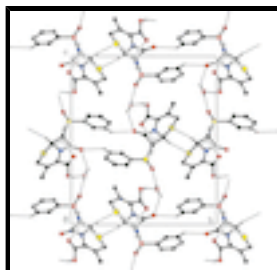


Fig. 2. The packing and hydrogen bonding of the title compound viewed down *a* axis. Hydrogen atoms not involved in hydrogen bonding have been omitted for clarity.

7-Benzenesulfonamido-3-methyl-8-oxo-5-thia-1-azabicyclo[4.2.0]octa-2-ene- 2-carboxylic acid monohydrate

Crystal data

$C_{14}H_{14}N_2O_5S_2 \cdot H_2O$

$M_r = 372.43$

Orthorhombic, $P2_12_12_1$

Hall symbol: P 2ac 2ab

$a = 5.9535$ (7) Å

$b = 15.8248$ (19) Å

$c = 18.411$ (2) Å

$V = 1734.6$ (3) Å³

$Z = 4$

$F(000) = 776$

$D_x = 1.426$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 1402 reflections

$\theta = 2.8$ – 17.3°

$\mu = 0.34$ mm⁻¹

$T = 296$ K

Prismatic, light yellow

$0.20 \times 0.20 \times 0.20$ mm

Data collection

Bruker Kappa APEXII CCD area-detector diffractometer

Radiation source: sealed tube

graphite

φ and ω scans

10847 measured reflections

3572 independent reflections

1890 reflections with $I > 2\sigma(I)$

$R_{int} = 0.064$

$\theta_{max} = 26.8^\circ$, $\theta_{min} = 2.6^\circ$

$h = -4 \rightarrow 7$

$k = -12 \rightarrow 20$

$l = -15 \rightarrow 23$

Refinement

Refinement on F^2

Least-squares matrix: full

Hydrogen site location: inferred from neighbouring sites

H atoms treated by a mixture of independent and constrained refinement

$R[F^2 > 2\sigma(F^2)] = 0.055$	$w = 1/[\sigma^2(F_o^2) + (0.0407P)^2]$
$wR(F^2) = 0.109$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 0.97$	$(\Delta/\sigma)_{\max} < 0.001$
3572 reflections	$\Delta\rho_{\max} = 0.20 \text{ e } \text{\AA}^{-3}$
226 parameters	$\Delta\rho_{\min} = -0.25 \text{ e } \text{\AA}^{-3}$
3 restraints	Extinction correction: <i>SHELXL97</i> (Sheldrick, 2008), $F_c^* = kFc[1+0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$
Primary atom site location: structure-invariant direct methods	Extinction coefficient: 0.0083 (10)
Secondary atom site location: difference Fourier map	Absolute structure: Flack (1983), 1425 Friedel pairs
	Flack parameter: -0.07 (11)

Special details

Geometry. Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement on F^2 for ALL reflections except those flagged by the user for potential systematic errors. Weighted R -factors wR and all goodnesses of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The observed criterion of $F^2 > \sigma(F^2)$ is used only for calculating $-R$ -factor-obs *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.8308 (2)	0.30975 (7)	0.07531 (7)	0.0628 (5)
S2	1.06325 (17)	0.48679 (7)	-0.11299 (6)	0.0452 (4)
O1	0.1740 (7)	0.4774 (2)	0.21094 (17)	0.0880 (16)
O2	0.1461 (5)	0.51706 (18)	0.09585 (15)	0.0573 (11)
O3	0.6341 (5)	0.56743 (17)	0.05463 (18)	0.0719 (13)
O4	1.3013 (4)	0.48188 (19)	-0.11039 (15)	0.0595 (11)
O5	0.9420 (5)	0.44363 (17)	-0.16919 (15)	0.0554 (10)
N1	0.5130 (5)	0.42642 (19)	0.06042 (18)	0.0410 (11)
N2	0.9775 (5)	0.44881 (19)	-0.03638 (18)	0.0447 (12)
C1	0.6086 (8)	0.2712 (3)	0.1326 (3)	0.072 (2)
C2	0.4335 (7)	0.3333 (3)	0.1575 (2)	0.0500 (17)
C3	0.2901 (9)	0.2995 (3)	0.2184 (2)	0.077 (2)
C4	0.3988 (6)	0.4074 (3)	0.1247 (2)	0.0427 (16)
C5	0.2262 (7)	0.4725 (3)	0.1418 (3)	0.0507 (17)
C6	0.6407 (7)	0.3665 (2)	0.0176 (2)	0.0500 (17)
C7	0.6267 (6)	0.4950 (3)	0.0341 (2)	0.0480 (16)
C8	0.7372 (7)	0.4416 (3)	-0.0250 (2)	0.0470 (17)
C9	0.9814 (6)	0.5938 (2)	-0.1135 (2)	0.0430 (14)
C10	1.1108 (7)	0.6515 (3)	-0.0765 (2)	0.0587 (17)
C11	1.0450 (10)	0.7349 (3)	-0.0734 (3)	0.074 (2)

supplementary materials

C12	0.8513 (10)	0.7598 (3)	-0.1058 (3)	0.069 (2)
C13	0.7206 (8)	0.7021 (3)	-0.1421 (3)	0.0683 (19)
C14	0.7853 (7)	0.6178 (3)	-0.1457 (3)	0.0577 (19)
OW1	0.8355 (9)	0.5874 (3)	0.2110 (3)	0.128 (2)
H1	0.07960	0.51450	0.21690	0.1320*
H1A	0.53260	0.22610	0.10690	0.0860*
H1B	0.67690	0.24650	0.17550	0.0860*
H2	1.07130	0.43340	-0.00340	0.0530*
H3A	0.17150	0.33880	0.22870	0.1160*
H3B	0.38080	0.29200	0.26100	0.1160*
H3C	0.22680	0.24620	0.20430	0.1160*
H6	0.54600	0.32940	-0.01210	0.0600*
H8	0.65710	0.44840	-0.07110	0.0560*
H10	1.24240	0.63430	-0.05370	0.0710*
H11	1.13350	0.77410	-0.04910	0.0890*
H12	0.80690	0.81610	-0.10330	0.0830*
H13	0.58820	0.71950	-0.16430	0.0820*
H14	0.69620	0.57840	-0.16970	0.0690*
HW1	0.706 (6)	0.587 (5)	0.228 (3)	0.1920*
HW2	0.819 (12)	0.614 (4)	0.172 (2)	0.1920*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0590 (8)	0.0401 (6)	0.0892 (10)	0.0106 (6)	0.0078 (8)	0.0044 (6)
S2	0.0366 (6)	0.0512 (7)	0.0479 (7)	0.0031 (5)	0.0001 (6)	0.0012 (6)
O1	0.104 (3)	0.110 (3)	0.050 (2)	0.032 (3)	0.014 (2)	-0.012 (2)
O2	0.0538 (19)	0.0611 (19)	0.057 (2)	0.0140 (18)	0.0046 (17)	-0.0017 (17)
O3	0.062 (2)	0.0368 (17)	0.117 (3)	0.0035 (17)	0.029 (2)	-0.0053 (18)
O4	0.0328 (16)	0.080 (2)	0.0656 (19)	0.0071 (16)	0.0052 (15)	0.0055 (18)
O5	0.0591 (18)	0.0585 (17)	0.0485 (19)	-0.0021 (16)	-0.0076 (17)	-0.0051 (15)
N1	0.0398 (19)	0.0353 (18)	0.048 (2)	0.0017 (17)	0.0067 (18)	-0.0016 (17)
N2	0.029 (2)	0.054 (2)	0.051 (2)	0.0090 (16)	0.0011 (16)	0.0119 (18)
C1	0.077 (4)	0.048 (3)	0.091 (4)	-0.003 (3)	-0.002 (3)	0.012 (3)
C2	0.043 (3)	0.057 (3)	0.050 (3)	-0.010 (3)	0.000 (2)	0.007 (2)
C3	0.080 (4)	0.086 (4)	0.066 (3)	-0.013 (3)	-0.005 (3)	0.017 (3)
C4	0.038 (2)	0.050 (3)	0.040 (3)	-0.005 (2)	-0.001 (2)	-0.002 (2)
C5	0.040 (3)	0.063 (3)	0.049 (3)	-0.005 (2)	0.004 (2)	-0.012 (3)
C6	0.048 (3)	0.046 (3)	0.056 (3)	-0.010 (2)	0.000 (2)	-0.011 (2)
C7	0.041 (3)	0.039 (2)	0.064 (3)	0.005 (2)	0.003 (2)	0.010 (2)
C8	0.037 (3)	0.054 (3)	0.050 (3)	0.006 (2)	-0.003 (2)	0.004 (2)
C9	0.037 (2)	0.044 (2)	0.048 (3)	-0.005 (2)	0.004 (2)	0.007 (2)
C10	0.052 (3)	0.058 (3)	0.066 (3)	-0.005 (3)	-0.012 (3)	0.001 (3)
C11	0.094 (4)	0.046 (3)	0.083 (4)	-0.015 (3)	-0.020 (4)	-0.004 (3)
C12	0.082 (4)	0.044 (3)	0.081 (4)	0.001 (3)	0.009 (4)	0.000 (3)
C13	0.053 (3)	0.062 (3)	0.090 (4)	0.012 (3)	-0.012 (3)	0.013 (3)
C14	0.047 (3)	0.048 (3)	0.078 (4)	-0.004 (2)	-0.004 (3)	0.002 (2)
OW1	0.152 (4)	0.104 (3)	0.129 (4)	0.035 (3)	0.085 (4)	0.000 (3)

Geometric parameters (Å, °)

S1—C1	1.799 (5)	C6—C8	1.535 (6)
S1—C6	1.793 (4)	C7—C8	1.527 (6)
S2—O4	1.420 (3)	C9—C14	1.363 (6)
S2—O5	1.435 (3)	C9—C10	1.375 (6)
S2—N2	1.616 (3)	C10—C11	1.378 (7)
S2—C9	1.762 (3)	C11—C12	1.357 (8)
O1—C5	1.313 (6)	C12—C13	1.373 (7)
O2—C5	1.200 (6)	C13—C14	1.390 (7)
O3—C7	1.208 (5)	C1—H1B	0.9700
O1—H1	0.8200	C1—H1A	0.9700
OW1—HW2	0.84 (5)	C3—H3C	0.9600
OW1—HW1	0.83 (4)	C3—H3A	0.9600
N1—C4	1.398 (5)	C3—H3B	0.9600
N1—C7	1.368 (5)	C6—H6	0.9800
N1—C6	1.449 (5)	C8—H8	0.9800
N2—C8	1.450 (5)	C10—H10	0.9300
N2—H2	0.8600	C11—H11	0.9300
C1—C2	1.504 (7)	C12—H12	0.9300
C2—C4	1.335 (6)	C13—H13	0.9300
C2—C3	1.507 (6)	C14—H14	0.9300
C4—C5	1.489 (6)		
S1...N2	3.136 (3)	C5...OW1 ⁱⁱⁱ	3.216 (7)
S1...C5 ⁱ	3.698 (5)	C7...O4 ⁱⁱⁱ	3.297 (5)
S1...H2	2.8200	C7...O2 ⁱ	3.313 (5)
S1...H6 ⁱⁱ	2.8000	C7...O2	3.099 (5)
O1...OW1 ⁱⁱⁱ	2.663 (6)	C8...C14	3.577 (7)
O1...C3	2.902 (6)	C8...O4 ⁱⁱⁱ	3.100 (5)
OW1...O2 ⁱ	3.025 (6)	C11...O3 ^{vii}	3.192 (6)
OW1...O3	3.135 (6)	C12...O3 ^{vii}	3.346 (6)
OW1...C5 ⁱ	3.216 (7)	C14...C8	3.577 (7)
OW1...O5 ^{iv}	2.799 (6)	C4...H2 ⁱⁱⁱ	3.0900
OW1...O1 ⁱ	2.663 (6)	C5...H3A	2.6700
O2...N1	2.693 (4)	C5...H2 ⁱⁱⁱ	2.8900
O2...C7 ⁱⁱⁱ	3.313 (5)	C9...H3B ^{vi}	3.0500
O2...C7	3.099 (5)	C12...H3B ^{vi}	3.0400
O2...OW1 ⁱⁱⁱ	3.025 (6)	C13...H3B ^{vi}	2.9700
O2...O3	3.107 (4)	C14...H3B ^{vi}	2.9900
O2...N2 ⁱⁱⁱ	2.846 (4)	H1...HW2 ⁱⁱⁱ	2.3600
O3...O2	3.107 (4)	H1...OW1 ⁱⁱⁱ	1.8600
O3...OW1	3.135 (6)	H1...HW1 ⁱⁱⁱ	2.5100
O3...C5	3.276 (6)	HW1...O5 ^{iv}	2.14 (5)
O3...C12 ^v	3.346 (6)	HW1...H1 ⁱ	2.5100

supplementary materials

O3...N2	3.242 (4)	H1A...H3C	2.5800
O3...C11 ^v	3.192 (6)	H1B...H3B	2.4700
O4...C7 ⁱ	3.297 (5)	H2...O2 ⁱ	2.3000
O4...C8 ⁱ	3.100 (5)	H2...N1 ⁱ	2.8800
O5...OW1 ^{vi}	2.799 (6)	H2...S1	2.8200
O1...H3A	2.2200	H2...C5 ⁱ	2.8900
OW1...H1 ⁱ	1.8600	H2...C4 ⁱ	3.0900
O2...HW2 ⁱⁱⁱ	2.85 (6)	HW2...O3	2.54 (5)
O2...H2 ⁱⁱⁱ	2.3000	HW2...O2 ⁱ	2.85 (6)
O2...H12 ^v	2.8100	HW2...H1 ⁱ	2.3600
O3...HW2	2.54 (5)	H3A...O1	2.2200
O3...H12 ^v	2.8300	H3A...C5	2.6700
O3...H11 ^v	2.5100	H3B...C12 ^{iv}	3.0400
O4...H10	2.6500	H3B...C14 ^{iv}	2.9900
O4...H8 ⁱ	2.3000	H3B...C9 ^{iv}	3.0500
O5...HW1 ^{vi}	2.14 (5)	H3B...C13 ^{iv}	2.9700
O5...H14	2.5900	H3B...H1B	2.4700
O5...H8	2.4800	H3C...H1A	2.5800
N1...O2	2.693 (4)	H6...S1 ^{viii}	2.8000
N2...O2 ⁱ	2.846 (4)	H8...O5	2.4800
N2...S1	3.136 (3)	H8...O4 ⁱⁱⁱ	2.3000
N2...O3	3.242 (4)	H10...O4	2.6500
N1...H2 ⁱⁱⁱ	2.8800	H11...O3 ^{vii}	2.5100
C3...O1	2.902 (6)	H12...O3 ^{vii}	2.8300
C5...O3	3.276 (6)	H12...O2 ^{vii}	2.8100
C5...S1 ⁱⁱⁱ	3.698 (5)	H14...O5	2.5900
C1—S1—C6	93.1 (2)	S2—C9—C14	120.5 (3)
O4—S2—O5	120.03 (18)	C10—C9—C14	120.7 (4)
O4—S2—N2	105.40 (17)	C9—C10—C11	119.8 (4)
O4—S2—C9	109.19 (18)	C10—C11—C12	120.1 (5)
O5—S2—N2	107.07 (17)	C11—C12—C13	120.2 (5)
O5—S2—C9	108.33 (17)	C12—C13—C14	120.3 (5)
N2—S2—C9	105.95 (17)	C9—C14—C13	118.9 (4)
C5—O1—H1	110.00	S1—C1—H1A	108.00
HW1—OW1—HW2	103 (6)	S1—C1—H1B	108.00
C6—N1—C7	93.8 (3)	C2—C1—H1B	108.00
C4—N1—C6	125.1 (3)	H1A—C1—H1B	107.00
C4—N1—C7	135.4 (3)	C2—C1—H1A	108.00
S2—N2—C8	117.8 (3)	C2—C3—H3B	109.00
C8—N2—H2	121.00	C2—C3—H3C	109.00
S2—N2—H2	121.00	H3A—C3—H3B	109.00
S1—C1—C2	117.8 (3)	H3A—C3—H3C	110.00
C1—C2—C4	122.9 (4)	H3B—C3—H3C	109.00
C3—C2—C4	124.1 (4)	C2—C3—H3A	109.00

C1—C2—C3	112.8 (4)	S1—C6—H6	113.00
N1—C4—C2	119.8 (4)	C8—C6—H6	113.00
N1—C4—C5	111.5 (4)	N1—C6—H6	113.00
C2—C4—C5	128.2 (4)	C6—C8—H8	110.00
O1—C5—C4	114.2 (4)	C7—C8—H8	110.00
O2—C5—C4	122.1 (5)	N2—C8—H8	110.00
O1—C5—O2	123.7 (4)	C11—C10—H10	120.00
S1—C6—N1	109.7 (3)	C9—C10—H10	120.00
S1—C6—C8	117.0 (3)	C10—C11—H11	120.00
N1—C6—C8	88.2 (3)	C12—C11—H11	120.00
N1—C7—C8	91.5 (3)	C13—C12—H12	120.00
O3—C7—N1	131.3 (4)	C11—C12—H12	120.00
O3—C7—C8	137.1 (4)	C12—C13—H13	120.00
N2—C8—C7	119.0 (3)	C14—C13—H13	120.00
C6—C8—C7	84.4 (3)	C9—C14—H14	121.00
N2—C8—C6	120.3 (4)	C13—C14—H14	121.00
S2—C9—C10	118.7 (3)		
C6—S1—C1—C2	-46.4 (4)	S1—C1—C2—C3	-166.4 (3)
C1—S1—C6—N1	57.8 (3)	S1—C1—C2—C4	18.8 (6)
C1—S1—C6—C8	156.1 (3)	C1—C2—C4—N1	6.2 (6)
O4—S2—N2—C8	-176.8 (3)	C1—C2—C4—C5	177.2 (4)
O5—S2—N2—C8	-47.9 (3)	C3—C2—C4—N1	-168.1 (4)
C9—S2—N2—C8	67.6 (3)	C3—C2—C4—C5	3.0 (7)
O4—S2—C9—C10	-32.7 (4)	N1—C4—C5—O1	-155.7 (4)
O4—S2—C9—C14	151.9 (4)	N1—C4—C5—O2	23.2 (6)
O5—S2—C9—C10	-165.0 (3)	C2—C4—C5—O1	32.6 (6)
O5—S2—C9—C14	19.5 (4)	C2—C4—C5—O2	-148.5 (5)
N2—S2—C9—C10	80.4 (3)	S1—C6—C8—N2	19.3 (5)
N2—S2—C9—C14	-95.1 (4)	S1—C6—C8—C7	-101.1 (3)
C6—N1—C4—C2	11.5 (6)	N1—C6—C8—N2	130.5 (4)
C6—N1—C4—C5	-161.0 (4)	N1—C6—C8—C7	10.1 (3)
C7—N1—C4—C2	-134.5 (5)	O3—C7—C8—N2	45.1 (7)
C7—N1—C4—C5	53.1 (6)	O3—C7—C8—C6	166.7 (5)
C4—N1—C6—S1	-49.9 (4)	N1—C7—C8—N2	-132.3 (4)
C4—N1—C6—C8	-168.0 (3)	N1—C7—C8—C6	-10.7 (3)
C7—N1—C6—S1	106.9 (3)	S2—C9—C10—C11	-177.2 (4)
C7—N1—C6—C8	-11.2 (3)	C14—C9—C10—C11	-1.7 (6)
C4—N1—C7—O3	-13.7 (7)	S2—C9—C14—C13	177.0 (4)
C4—N1—C7—C8	164.0 (4)	C10—C9—C14—C13	1.6 (7)
C6—N1—C7—O3	-166.4 (4)	C9—C10—C11—C12	1.0 (7)
C6—N1—C7—C8	11.3 (3)	C10—C11—C12—C13	-0.3 (8)
S2—N2—C8—C6	143.8 (3)	C11—C12—C13—C14	0.2 (8)
S2—N2—C8—C7	-114.9 (3)	C12—C13—C14—C9	-0.8 (8)

Symmetry codes: (i) $x+1, y, z$; (ii) $x+1/2, -y+1/2, -z$; (iii) $x-1, y, z$; (iv) $-x+3/2, -y+1, z+1/2$; (v) $x-1/2, -y+3/2, -z$; (vi) $-x+3/2, -y+1, z-1/2$; (vii) $x+1/2, -y+3/2, -z$; (viii) $x-1/2, -y+1/2, -z$.

Hydrogen-bond geometry ($\text{\AA}, ^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
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supplementary materials

O1—H1…OW1 ⁱⁱⁱ	0.82	1.86	2.663 (6)	166
OW1—HW1…O5 ^{iv}	0.83 (4)	2.14 (5)	2.799 (6)	136 (5)
N2—H2…S1	0.86	2.82	3.136 (3)	103
N2—H2…O2 ⁱ	0.86	2.30	2.846 (4)	122
OW1—HW2…O3	0.84 (5)	2.54 (5)	3.135 (6)	129 (5)
C3—H3A…O1	0.96	2.22	2.902 (6)	127
C6—H6…S1 ^{viii}	0.98	2.80	3.756 (4)	165
C8—H8…O4 ⁱⁱⁱ	0.98	2.30	3.100 (5)	138
C8—H8…O5	0.98	2.48	2.922 (5)	107
C11—H11…O3 ^{vii}	0.93	2.51	3.192 (6)	130
C14—H14…O5	0.93	2.59	2.942 (5)	103
C3—H3B…Cg3 ^{iv}	0.96	2.72	3.640 (5)	161

Symmetry codes: (iii) $x-1, y, z$; (iv) $-x+3/2, -y+1, z+1/2$; (i) $x+1, y, z$; (viii) $x-1/2, -y+1/2, -z$; (vii) $x+1/2, -y+3/2, -z$.

Fig. 1

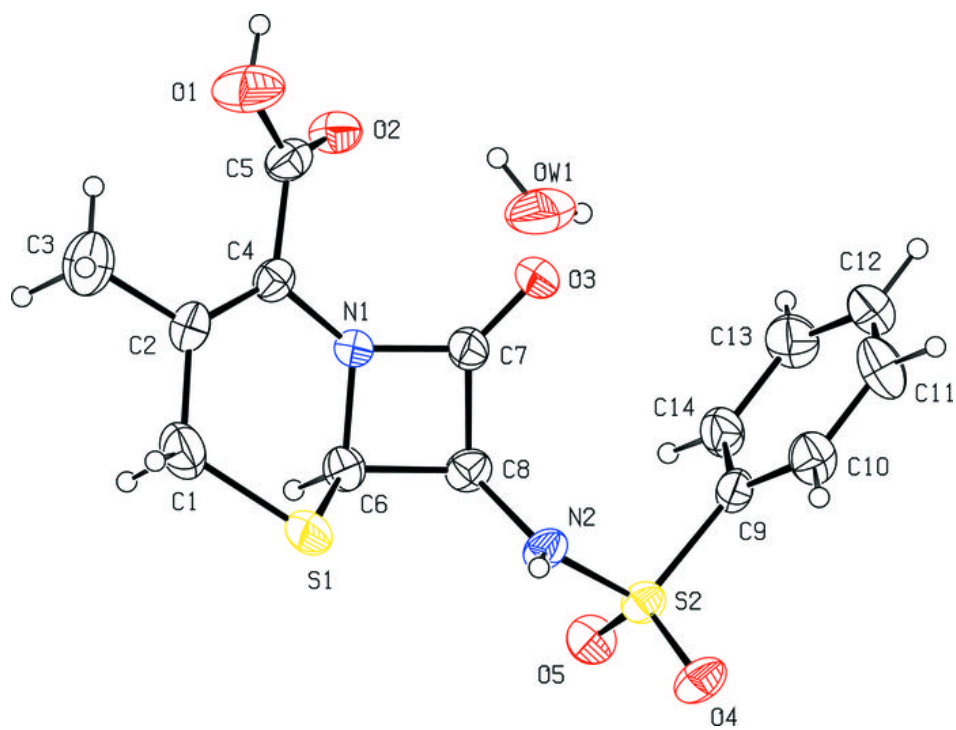


Fig. 2

